

STAKHEYEV, B.F., aspirant

Effect of feed grisin on the blood indices of fattening  
pigs. Veterinariia 40 no.3:55-57 Mr '63. (MIRA 17:1)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut zhivotno-  
vodstva.

STAKHAYEV, D.D.; TOBIAS, D.A., kandidat tekhnicheskikh nauk, retsenzent;  
TAURIL, G.E., inzhener, retsenzent; AVILOV, V.M., redaktor;  
MODEL', B.I., tekhnicheskii redaktor

[The assembly line in mass machine production] Potochnaia liniia  
v massovom mashinostroenii. Moskva, Gos. nauchno-tekhn. izd-vo  
mashinostroit. lit-ry, 1951 202 p. [Microfilm] (MLRA 10:1)  
(Machinery industry) (Assembly line methods)

STAKHEYEV, D. D.

Assembly-line methods

"Production line in mass machine construction." Sov. kniga no. 2, '52.

Monthly List of Russian Accessions, Library of Congress, July 1952. Unclassified.

STAKHEEV, D.D.

USSR/Miscellaneous - Machine Tools

Card : 1/1

Authors : Stakheev, D. D.

Title : About the deficiencies of present technological processes in machining gear wheels.

Periodical : Stan. i instr.<sup>25</sup>, 3, 5 - 9, Mar 1954

Abstract : A detailed analysis of the causes of gear wheel displacements is given and a method of broaching is described, which will insure the centricity of the holes. This method was developed at the Stalin Automobile Factory. Diagrams.

Institution : ....

Submitted : ....

32 (2)

SOV/112-57-5-10860

Translation from: Referativnyy zhurnal. Elektrotehnika, 1957, Nr 5, p 183 (USSR)

AUTHOR: Stakheyev, D. D.

TITLE: Experience of Automobile Plant imeni Stalin With Designing and Adopting Automation Means (Opyt raboty avtozavoda im. Stalina po konstruirovaniyu i vnedreniyu sredstv avtomatizatsii)

PERIODICAL: V sb.: Avtomatizatsiya tekhnol. protsessov v mashinostr. Obrabotka metallov rezaniyem i obshchiye vopr. avtomatizatsii. M., 1956, pp 225-239

ABSTRACT: Bibliographic entry.

Card 1/1

STAKHUYEV, D.D.

Tasks of the automobile industry in 1958. Avt. prom. no.1:1-2 Ja  
'58. (MIRA 11:2)

1. Gosplan SSSR.

(Automobile industry)

STAKHEYEV, D.D.

Automobile industry for the period 1959-1965. Avt.prom.  
no.1:1-2 Ja '59. (MIRA 12:1)

1. Gosplan SSSR.  
(Automobile industry)

STAKHEYEV, D.

New types of motor vehicles in 1959. Avt.transp. 37 no.3:36-37  
Mr '59. (MIRA 12:4)

1. Otdel mashinostroyeniya Gosplana SSSR.  
(Motor vehicles)



STAKHEYEV, D., inzh.

Development of the automobile industry. Avt.trans.  
38 no.8:25-26 Ag '60. (MIRA 13:8)  
(Automobile industry)

L 13626-65 AMD  
ACCESSION NR: AR4045864

S/0299/64/000/014/M023/M023

SOURCE: Ref. zh. Biologiya. Svodnyy tom, Abs. 14M151

AUTHOR: Stetsula, V. I.; Stalkeyev, I. A.

TITLE: Primary accretion of bone heterotransplants

CITED SOURCE: Sb. 3 Vses. konferentsiya po peresadke tkaney i organov, 1963. Yerevan, 1963, 459-460

TOPIC TAGS: accretion, bone, transplantation, compression  
arthrodesis, knee, dogs

TRANSLATION: In experiments on 31 dogs, the accretion of heterotransplants was studied under conditions of compression arthrodesis of the knee joint with the use of a device of G. A. Ilizarov. Spongy or compact bone lamina taken from calves or humans with a 10 to 17 mm thickness and preserved by freezing were used as heterotransplants. A roentgenological and histological analysis (after 16 to 360 days) showed primary bone accretion of the heterotransplant. The use of transplants with considerable

Card 1/2

L 13626-65  
ACCESSION NR: AR4045864

mechanical durability (thickness 30-40 mm) is recommended to prevent secondary resorption resulting from dynamic loading.

SUB CCDE: LS

ENCL: 00

Card 2/2

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<div style="display: flex; justify-content: space-between;"> <div> <p>COMMON ELEMENTS</p> <p>OPEN MATERIAL INDEX</p> </div> <div> <p>STAKHEYEV I. S.</p> <p>Processes and Properties of the</p> </div> <div> <p>9</p> </div> </div>																																																																																																																																	
<p>Ore treatment at the Uraznominidze concentrator of the Kazoloto Trust. I. S. Stakheyev, <i>Zolotaya Prom.</i> 12, No. 7, 23-4(1940).—The concentrator handles ores contg. small amts. of stibnite and also old slimes. The peculiarities of the scheme are the grinding and amalgamation in a water medium and inside amalgamation in grinding pans and mills equipped with special amalgamators. The Sb minerals produce rapid "exhaustion" of the solns. and this necessitates a large consumption (up to 1800 g./ton) of cyanide.</p> <p style="text-align: right;">B. Z. Kamich</p>																																																																																																																																	
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STAKHEYEV, I. S., ENGINEER

"Kinetics of Cyaniding Gold in the Presence of Oxidized Lead Minerals." Sub 10 Apr 47, Moscow Inst of Nonferrous Metals and Gold  
imeni M. I. Kalinin

*Cand Tech Sci*  
Dissertations presented for degrees in science and engineering in  
Moscow in 1947.

So: Sum. No. 257, 18 Apr 55

5.  
STAKHEEV, I., gornyy inzhener; FAYNER, I.

For a continuous work organization in stopes. Ugol' 29 no.6:44-48 Je '54.  
(MLRA 7:6)

1. Shakhta "Chernaya Gora" v Kuzbasse. (Coal mines and mining)

0.0000

1776

807/144-86-1-27/27

AUTHOR: Stakheyev, I. S.

TITLE: Scientific Chronicle. The Gold Industry of Ghana  
(Based on Materials of Mission)

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy. Tsvetnaya  
metallurgiya, 1960, Nr 1, pp 166-171 (USSR)

ABSTRACT: This article contains information on the gold industry  
of Ghana, Africa.

Card 1/1

UKRAINSKIY, M.A., st. nauchn. sotr.; MASKEVICH, M.M.; LODEYSHCHIKOV, V.V., kand. tekhn. nauk; SKOBEYEV, I.K., prof., doktor tekhn. nauk; STAKHEYEV, I.S., kand. tekhn. nauk; KULIKOV, A.V., kand. tekhn. nauk; KULIKOVA, S.Ya., kand. geol.-miner. nauk; FOKROVSKIY, L.A.; ALEKSANDROVA, N.N.; YELANSKIY, A.N., st. nauchn. sotr.; TROKSKAYA, Z.I.; BANDENOK, L.I., nauchn. sotr.; VERIGO, K.N.; TEMKO, V.P., red.

[Gold mining industry in capitalist countries; technical and economic survey] Zolotodobyvaiushchaia promyshlennost' kapitalisticheskikh stran; tekhniko-ekonomicheskii obzor. Moskva, 1963. 337 p. (MIRA 17:6)

1. TSentral'nyy nauchno-issledovatel'skiy institut informatsii i tekhniko-ekonomicheskikh issledovaniy tsvetnoy metallugii.
2. TSentral'nyy nauchno-issledovatel'skiy institut informatsii i tekhniko-ekonomicheskikh issledovaniy tsvetnoy metallurgii (for Ukrainskiy, Yelanskiy, Verigo).



STANLEY, I.V., Cand Tech Sci --(diss) "Study of masquerade-<sup>during</sup>  
aeration of yeast in an apparatus of <sup>the film</sup> ~~pollen~~ type." Minsk, 1958.  
25 pp (Min of Higher Education USSR. Belorussian Polytech Inst in I.V. Sta-  
lin. Chair of Heat and Gas Supply and Ventilation), 100 copies (KL,26-58,  
112)

STAKHEYEV, I.V.; inzh.

Investigating the mass transfer during the production of baking  
yeast under conditions of film diffusion. Sbor. nauch. rab. Bel.  
politekh. inst. no. 69:29-50 '58. (MIRA 12:7)  
(Mass transfer) (Yeast)

STAKHEYEV, I.V.

Determining the coefficient of moisture exchange during the aeration  
of a yeast suspension in a film-type apparatus. Sbor.nauch.trud.  
Bel.politekh.inst. no.87:39-44 '59. (MIRA 14:4)  
(Yeast)

KHAZANOV, Y.I.; STANKEYEVA, S.A.; KOS'MINA, G.V.

Interaction between sodium aluminate and dicalcium silicate.  
Zhur.prikl.khim. 38 no.6:1381-1383 Je '65.

(MIRA 18:10)

S/137/62/000/004/191/201  
A154/A101

AUTHOR: Stakheyev, Yu. I.

TITLE: Use of photoelectric registration methods in the spectral analysis of ores

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 4, 1962, 2, abstract 4K10 ("Nekotoryye vopr. emission. i molekulyarn. spektroskopii", Krasnoyarsk. 1960, 115-120)

TEXT: A method has been worked out for determining Li and In in ores. A spectral instrument based on CT-7 (ST-7) stylometer, has been built for analyzing Li, that permits subdividing the analyzed light into 2 beams, - an analytic line and a comparison line. Assay is introduced into discharger by air jet. The arc's current is 15a. The error in one measurement is 2.5%. In is determined with the aid of a spectral instrument fitted with KC-55 (KS-55) glass optics. When determining small amounts of In, the background of continuous spectrum was taken into account with the aid of a registering circuit in a photoelectric apparatus. In this circuit the continuously-varying light intensity

Card 1/2

Use of photoelectric registration ...

S/137/62/000/004/191/201  
A154/A101

is transduced into a discretely changing number of pulses, the number of which can serve as a criterion for the registered intensity of light.

L. Vorob'yeva

[Abstracter's note: Complete translation]

Card 2/2

STAKHEYEV, Yu.I.

Atomic emission spectrum analysis from the standpoint of the  
theory of information. Zav.lab. 28 no.7:831-835 '62. (MIRA 15:6)  
(Spectrum analysis)

LOSEV, N.F.; SMAGUNOVA, A.N.; STAKHEYEV, Yu.I.

Modern methods of X-ray spectral fluorescence analysis (survey).  
Zav.lab. 30 no.4:420-425 '64. (MIRA 17:4)



RAYKHRAUM, Ya.D.; STAKHEYEV, Yu.I.

Scintillation-spectral method of mineralogical analysis. Zhur. anal.  
khim. 20 no.3:299-304 '65. (MIRA 18:5)

1. Institut geokhimii i analiticheskoy khimii imeni Vernadskogo AN  
SSSR, Moskva.

KHAZANOV, Ye.I.; KUZ'MINA, G.V.; STAKHEYEVA, S.A.; SHUL'TS, B.V.

Changes in the phase composition of clays during heating in a neutral atmosphere in the presence of a solid reducing agent. Trudy Vost.-Sib. fil. AN SSSR no.43:69-76 '62. (MIRA 16:3)

(Aluminum oxide)

(Clay)

(Phase rule and equilibrium)

S/897/62/000/043/001/001  
B117/B186

AUTHORS: Khazanov, Ye. I., Saionova, Ye. G., Stakheyeva, S. A.,  
Kuz'mina, G. V.

TITLE: Reaction of aluminum carbide with magnesium oxide

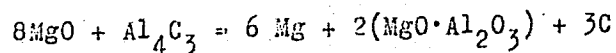
SOURCE: Akademiya nauk SSSR. Vostochno-Sibirskiy filial. Trudy. no.43.  
1962. Syr'yevyye resursy legkikh metalliv Vostochnoy Sibiri.  
v. 4, 112-128

TEXT: The reduction of magnesium oxide with aluminum carbide was studied both in theory and practice. The carbide was produced by heating metallic aluminum and electrode graphite, parts by weight ratio 2 : 1, for 3 hrs in the presence of 5% cryolite at 1350 - 1400°C in a hydrogen stream (5 l/hr). After treatment of the reaction products with 0.5 N HCl solution in the cold and increasing the hydrogen stream to 10 l/hr, the  $Al_4C_3$  content rose from 50% to 73 - 90%. Magnesium oxide reacted rapidly with  $Al_4C_3$  on heating in vacuo in a special apparatus at a temperature as low as 900°C, forming metallic magnesium, carbon black, and spinel:

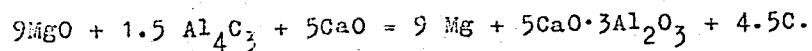
Card 1/2

Reaction of aluminum carbide with ...

S/897/62/000/043/001/001  
B117/B186



The magnesium yield, being ~7 - 10% increased with elevated temperatures, reaching 81-86% at 1200-1300°C. In the presence of calcium oxide  $\text{MgO} + \text{Al}_4\text{C}_3$  yielded pentacalcium trialuminate:



The magnesium yield was shown to increase by an excess of reducing agent,  $\text{MgO}$ , or  $\text{CaO}$ . An addition of calcium fluoride accelerated the reaction between  $\text{MgO}$  and  $\text{Al}_4\text{C}_3$ . There are 8 figures and 8 tables.

Card 2/2

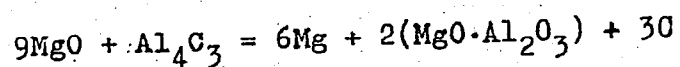
S/080/63/036/002/002/019  
D403/D307

AUTHORS: Khazanov, Ye. I., Safonova, Ye. G., Stakheyeva, S. A.  
and Kuzmina, G. V.

TITLE: The interaction of aluminum carbide with magnesium  
oxide

PERIODICAL: Zhurnal prikladnoy khimii, v. 36, no. 2, 1963, 251-263

TEXT: A brief review is first given of the physical and chemical properties of  $Al_4C_3$ . In the present work the authors prepared this compound by a number of methods, the best being the interaction of Al with electrode graphite, in the presence of cryolite, at 1350 - 1400°C, over 3 hours, under  $H_2$ . The mixture was then treated with HCl. The product was reacted with pure MgO, under vacuum, at 800 - 1300°C. The reaction

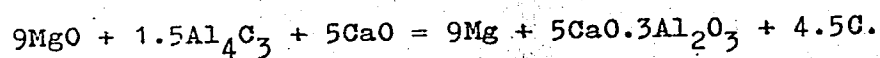


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D403/D307

The interaction of ...

proceeded appreciably at 800°C, fairly rapidly at 900°C (yield of Mg ~7 - 10%), and rapidly at 1200 - 1300°C (81 - 86% Mg). Small additions of CaF<sub>2</sub> accelerated the reaction. The products were confirmed by petrographic, chemical and x ray analyses. In the presence of lime, the reaction was found to be



There are 6 figures and 8 tables.

SUBMITTED: July 22, 1961

Card 2/2

Proteolytic enzymes of the soybean. E. D. Stakhovskaya, Kayerzheva and E. I. Oleinikova. *Biokhimiya* 17:321-324 (German 329-330) (1938).—Exts. of resting and germinating soybeans are equally active in reducing  $\eta$  of aq. gelatin, but while the exts. from the resting seeds show little or no proteolytic activity, those from the germinating seeds are much more active. During germination, the proteases of the seeds remain relatively const., but the peptidases increase. With peptone as substrate, the activity of the protease is optimum at pH 7.0, and with gelatin and collagen (I) at 7.2-7.4. The small activity at acid pH and the absence of any activating action by KCN and  $\text{Na}_2\text{S}$  indicate a low content of enthepsin (sepd. by adsorption on kaolin). Notwithstanding the specificity of plant proteases, the proteolytic enzymes of the germinating beans rapidly dissolve powd. I.

B. C. A.

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

PROCESS AND PROPERTY NOTES																									
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<p><i>Fission of animal proteins by the proteases of Aspergillus oryzae.</i> K. D. Stakhovna-Kavergina and E. I. Oleinikova. <i>Biozhivnaya</i> 1, 331-41 (in German 342) (1938). Exts. made from <i>A. oryzae</i>, grown on a medium prepd. from defatted soybean meal, hydrolyzed gelatin and collagen (optimum pH 7.2) but not elastin and keratin. Addn. of 0.1% of Na<sub>2</sub>SO<sub>4</sub> or (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> slightly increased the activity while greater concns. were inhibitory. By pptn. with EtOH, preps. were obtained almost equal in activity to pancreatin from ox pancreas. B. C. A.</p>																									
<p>ASB-SEA METALLURGICAL LITERATURE CLASSIFICATION</p>																									
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25

The chemical structure of wool keratin. I. Fractionation of keratin. V. Stakheva-Kavranova and N. P. Gavrilov. *Bull. Acad. Sci. USSR Div. Chem. Sci.* 19-27 (1937).—Wool of 30-40  $\mu$  thickness was immersed for 36-48 hrs. in a 0.1% soda soln. A mist of toluene and pancreatin activated by antecobalamin was added and digestion at 35-37° was continued for 3-30 days, the soln. being removed every 3-6 days. The keratin was thus sep'd. into 2 fractions, (1) the stable keratin of the cells which forms about 90% of the wool. It contains a higher percentage of N, S, arginine and cystine than wool; and (2) the intercellular substance which represents about 10% of the wool. This is disintegrated by pancreatin and is therefore not a typical keratin. It contains less N and S than wool and lacks the stability of keratin towards trypsin. Its compn. is characterized by a low content of diamino acids and a high content of monamino acids. Bibliography. M. M. Flannery

Lab. Org. Chem. in N. D. Zelinsky. Moscow State U.

ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION

STAKHEYEVA-KAVERZNEVA, Ye. D.

"On the Fermentability of the Simplest Diketopiperazines," Biokhim, 4, No. 2, 1939.  
From the Laboratory of Protein Chemistry, VIEM. Moscow, -1939-.

1ST AND 2ND ORDERS																																																																													
PROCESSES AND PROPERTIES INDEX																																																																													
<p><i>Ca</i></p> <p><b>Estimation of leucine and valine by the Fromageot and Heits method.</b> H. D. Makhsheva-Kavrinova. <i>Biochimie</i> 8, 513-20 (1940). The method of Fromageot and Heits (C. A. 33, 9201), which is based on the oxidation of valine and leucine by <math>\text{CrO}_3</math> to <math>\text{Me}_2\text{CO}</math> and coupling with salicylaldehyde, is accurate when only one of the amino acids is present in pure form. The method cannot be applied when both valine and leucine are present, or in protein hydrolyzates, even after fractionation of the Cu salts.</p> <p>H. Priestley</p>																																																																													
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STAKHEYEVA-KAVERZNEVA, YE. D.

"I. A Study of the Conditions of Hydrolysis of Serum Albumin," Zhur. Obshch. Khim.,  
13, No. 6, 1943. Lab. Chem. Albumin, All-Union Inst. Exptl., Med. im. A. M. Gor'k,  
-1942-. 6.402

STAKHOYEVA-KAVORZNEV, E. D.

"II. The Composition of Amino-Acids in Blood Albumins"., Stakhoyeva-Kavorznev, E. D.  
(p. 423)

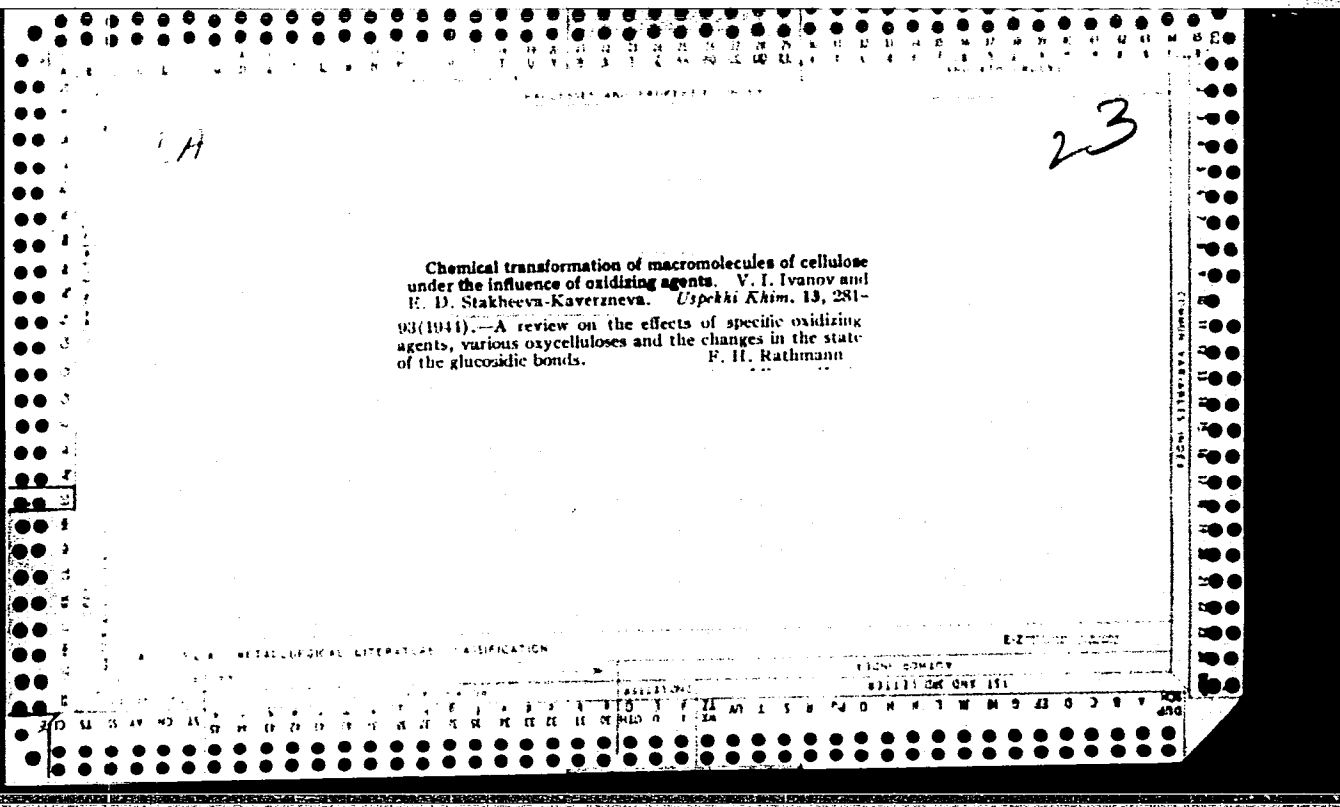
SO: Journal of General Chemistry (Zhurnal Obshchei Khimii) 1943, Volume 13, no. 6.

STAKHEYEVA-KAVERZNEV, E. D.

"III. Aldehydes in the Acid Hydrolysis of Serum Albumin". Stakheyeva-Kaverznev, E. D.  
(p. 427)

SO: Journal of General Chemistry (Zhurnal Obshchei Khimii) 1943, Volume 13, no. 6.

1ST AND 2ND ORDERS																										3RD AND 4TH ORDERS																									
PROCESSES AND PROPERTIES INDEX																																																			
<p>CA</p> <p>Chemical nature of blood proteins. IV. Hydroxy-amino acids in the serum proteins of blood. E. D. Stakheeva-Kavetskaia. <i>J. Gen. Chem. (U.S.S.R.)</i> 13, 736-40 (1943) (English summary); cf. C. A. 38, 333P. — The content of hydroxyamino acids in serum globulins exceeds by a factor of two their content in the albumins. Both fractions of proteins contain many unidentified amino acids. Electrophoretic sepn. of amino acids shows that the hydroxyamino acids are present in all 3 major fractions. It is suspected that hydroxylysine is present among the diamino acids, hydroxyvaline among the monoamino-monocarboxylic acids, and hydroxyglutamic acid among the dicarboxylic acids, in amts. greater than obtainable by direct sepn. methods. It is shown that in the hydrolyzates of globulins there are unknown unstable N compds., which are absent in albumins. It is supposed that hydroxyamino acids and other unstable compds. may play a role in the formation of the antibodies. G. M. K.</p> <p>Lab, Chemistry of Albumins, Inst. Exptl. Medicine in L. M. Gorking</p>																																																			
<p>COMMON ELEMENTS</p> <p>COMMON VARIANTS INDEX</p> <p>OPEN</p> <p>MATERIALS INDEX</p> <p>ASM-SLA METALLURGICAL LITERATURE CLASSIFICATION</p> <p>1ST AND 2ND ORDERS</p> <p>3RD AND 4TH ORDERS</p> <p>1ST AND 2ND ORDERS</p> <p>3RD AND 4TH ORDERS</p>																																																			





COMMON ELEMENTS		COMMON RARE EARTH ELEMENTS		COMMON METALS		COMMON NON-METALS		COMMON GASES		COMMON LIQUIDS		COMMON SOLIDS		COMMON COMPOUNDS		COMMON MIXTURES		COMMON ALLOYS		COMMON POLYMERS		COMMON COMPOUNDS		COMMON MIXTURES		COMMON ALLOYS		COMMON POLYMERS	
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PROCESSES AND PROPERTIES INDEX

Chemical conversion of macromolecules of cellulose under the influence of oxidizing agents. I. The stability of the glucose bonds in cellulose. E. D. Stakhoeva-Kavtzeva and V. I. Ivanov. *Bull. acad. sci. U.R.S.S., Chem. sci. chim.* 1945, 683-8 (in English, 688); cf. C.A.B. 39, 17557. Monotriyl cellulose ethers, in which only the primary OH is substituted, are not oxidized in cupric ammonium suspension or in pyridine soln. by atm. O<sub>2</sub>. This indicates that oxidation of cellulose begins by conversion of the primary OH to aldehyde, which isomerizes with rupture of the cellulose chain. H. M. Leicester

ASTM-SLA METALLURGICAL LITERATURE CLASSIFICATION

STANDARD SYMBOL	STANDARD SYMBOL
STANDARD SYMBOL	STANDARD SYMBOL

CA

Chemical changes which take place in the cellulose macromolecule in copper-ammonia solution under the influence of atmospheric oxygen. V. I. Ivanov and E. D. Stakheva-Kaverzina. *Compt. rend. acad. sci. U.S.S.R.* 48, 406-8 (1968) (in English).—Passage of air through a 2-3% soln. of cellulose (I) in cuprammonium hydroxide at 18° for 30-90 days transforms 8% of the I into low-mol. decompn. products; at 30°, 25% of the I is decomposed. The decompn. products of I and those obtained on hydrolysis of the oxidized I by 0.2 N HCl were (in mols. per no. of glucose radicals in the I): formic acid 1:2180, 1:182; acetic acid 1:127, 1:98; oxalic acid, traces in each; glycolic acid, none, traces; erythronic acid, none in each; arabonic acid, — and present; gluconic acid, — and present; glucuronic acid (II), 1:61 in each; formaldehyde, absent, present; acetaldehyde, absent in each; furfural, present in each; CO<sub>2</sub> absent in first products. From the yield of II, it follows that 1 glucose radical of every 30 in the I is oxidized, 50% of these oxidized radicals passing into soln. Such mild oxidation decreases the viscosity of the I soln. nearly to that of the original solvent. The lower aliphatic acids were produced by secondary reactions. The product and their ratios class the oxidation process as a reaction which begins with the oxidation of the 6th C atom in the glucose radical. The data do not support the oxidation scheme of Staudinger (*C.A.* 33, 9291<sup>1</sup>). The mechanism of the decompn. of the I macromol. in alk. soln. under the action of oxidizing agents is discussed. C. J. West

23

CA  
Synthesis of 6-deoxycellulose. E. D. Stakheeva, K. V. Zvezdina, A. I. Ivanov, and A. S. Salova. *Zh. Obshch. Khim.* 1949, 369-78. 6-Deoxycellulose was prepd. via tosylcellulose, isotosylcellulose, unsatd. desoxycellulose; the products contained 0.35-0.55 moles reduced primary HO groups per glucose residue. Linters (1 g.) reprecipd. from cuprammonium soln., washed and dried by 3 changes of pyridine (3 days), treated in 25 ml. pyridine with 22.8 g. *p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>2</sub>Cl in pyridine and let stand at 15° gave a series of products; for substantially mono-substitution the reaction must be stopped before a homogeneous mass forms (10-20 hrs.), by treatment with aq. Me<sub>2</sub>CO at 0°; washing and extrn. with Me<sub>2</sub>CO or MeOH yielded 7 g. powdery product contg. 0.8-0.82 mole of ester groups on primary OH groups and 0.2-0.18 on secondary HO groups; for complete substitution of primary HO it is necessary to continue the reaction until total esterification reaches 1.5-1.88 ester moles per glucose unit, at which point the products become sol. in pyridine and Me<sub>2</sub>CO. The tosyl deriv. (0.8 g.), 0.8 g. NaI and 10 ml. dry Me<sub>2</sub>CO, heated in a sealed tube 1-4 hrs. at 110-20°, then washed with H<sub>2</sub>O give 80-90% *isolo* deriv., yellowish powder, which either is insol. in AcOH, EtOH, and dioxane, if the tosyl content is about 0.16 mole per 2 glucose units with 0.75 atom I, or insol. in AcOH and EtOH when the compn. is approx. 1:1:1 (from tosyl deriv. contg. 1.88 tosyl residues per 1.12

glucose). The tosyl group was best removed by shaking 1 g. of isotosylcellulose with 80-100 ml. dry piperidine, letting stand 3 days, filtering, dilg. with water, centrifuging, and reprecipg. the product from 5% AcOH by addn. of alkali to neutrality; the resulting unsatd. desoxycellulose is an amorphous, colorless powder, rapidly decolorizing KMnO<sub>4</sub> and Br; 2 analyzed products gave the following values: (C<sub>6</sub>H<sub>7</sub>O<sub>2</sub>)<sub>n</sub> and (C<sub>6</sub>H<sub>7</sub>O<sub>2</sub>)<sub>n</sub>(OH)<sub>m</sub> (tosyl); (C<sub>6</sub>H<sub>7</sub>O<sub>2</sub>)<sub>n</sub> and (C<sub>6</sub>H<sub>7</sub>O<sub>2</sub>)<sub>n</sub>(OH)<sub>m</sub> (as above) 0.1, 0.6, 2.10, 0.15, 0.014, 0.23, resp.; the products thus had 0.80 to 0.5 double bond per glucose unit. Reduction by Zn-AcOH or catalytic reduction by Pd in AcOH or dioxan proceeded readily; in the second method destructive hydrogenation appeared to occur also, giving low-mol. products. Reduction of isotosylcellulose by Na-piperidine gave a product still contg. 0.4 double bond per glucose; Zn-AcOH reduction of the isotosyl deriv., followed by Na-Hg detosylation, gave a low-mol. product contg. 0.20 tosyl unit per 1.86 glucose units and yielding

only 67.7% rhamnose; catalytic reduction of unsatd. desoxycellulose by Pd-BaSO<sub>4</sub> in AcOH left 0.1 double bond per glucose unit, with some residual tosyl, iodine, and piperidine groups, the product again being of low mol. wt.

G. M. Kosolapoff

Ketone groups in oxidized molecule of cellulose. E. D. Stukhcheva-Kaveranova. *Doklady Akad. Nauk SSSR* 68, 865-7 (1940). "Oxycellulose" prepd. by different procedures was analyzed for total carbonyl content by the hydroxylamine method, while the aldehyde detn. by controlled iodometric procedure (Martin, *et al.*, *C.I.* 36, 691<sup>1</sup>), gave by difference the content of ketonic CO groups. All "oxycelluloses" except that prepd. by NaIO<sub>4</sub> had some keto groups. The following results (in %) were obtained: purified cotton 0.04% CO<sub>2</sub>H and 0.01% keto groups, cotton oxidized by HIOCl at pH 4.6 0.21 and 0.34, oxidized at pH 6.8, 0.21 and 0.22, oxidized at pH 11 0.18 and 0.21, oxidized moderately at pH 9, 0.34 and 0.40, oxidized extensively at pH 9 1.2 and 0.62, oxidized by NO<sub>2</sub> 1.63 and 0.56, oxidized by H<sub>2</sub>O<sub>2</sub> in 8% NaOH 0.22 and 0.14, same in 2% NaOH 0.21 and 0.09, oxidized by H<sub>2</sub>O<sub>2</sub>-FeSO<sub>4</sub> in neutral medium 0.32 and 0.2, oxidized by neutral NaIO<sub>4</sub> soln. 2.66 and nearly 0% (traces), oxidized by CrO<sub>3</sub> in acid soln. 1.12 and 0.4, resp.

G. M. Kosolapoff

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Chemical changes of cellulose macromolecule under action of oxidizing agents. IV. Lactone links in the macromolecule of oxycellulose and their effect on determination of carbonyl groups by condensation with hydroxylamine. E. D. Stakheeva-Kavetznaya and A. S. Sakova. *Izv. Akad. Nauk S.S.S.R., Otdel. Khim. Nauk* 1951, 782-90; cf. C.I. 40, 5211; 44, 1257. Oxycellulose specimens prepd. by various means of oxidation: NaOCl, peroxide,  $H_2O_2$ , all oxidation in cuprammonium soln., gave a pink color when treated with hydroxylamine (at pH 7-8.5, washed with 0.1% HCl, and treated with a few drops of FeCl<sub>3</sub> soln.); this indicates ester linkage in all such specimens, such as would be afforded by lactone formation. If the oxycellulose is pretreated with 0.05 N NaOH the reaction does not appear. The formation of hydroxamic acids (color test above) occurs slowly even at pH 5-6 but not at pH 3. Reduction of the oxycellulose thus treated with hydroxylamine either by Cu and  $H_2O$ , Ca,  $H_2O$ , and dioxane, Zn and HCl, or electrolytically (all in the cold) gave products contg. primary  $NH_2$  groups which confirms the presence of ketone groups in oxycellulose.  $NH_3$  was evolved during reduction, indicating the decompn. of acid amides formed from reduction of hydroxamic acid groups. Since the lactone groups react with hydroxylamine under conditions that are generally used for detn. of carbonyl groups by this reagent, the results ob-

tained are higher than theoretical. V. Proof of existence of  $\alpha$ -hydroxymonoketone groups in oxycellulose. E. D. Stakheeva-Kavetznaya. *Ibid.* 791-4. Oxycellulose specimens oxidized by NaOCl,  $H_2O_2$ ,  $NO_2$ , or atm.  $O_2$  in cuprammonium soln., possess  $CH(OH)C(=O)$  groups, as shown by treatment with phosphotungstic reagent.  $HIO_4$  does not produce such groups in its reaction with cellulose. The hydroxyketo groups isomerize to enediol groups only at pH above 10; hence they do not interfere with the iodometric detn. of aldehyde groups. The blue color test (Benedict, C.I. 16, 2521) was checked on a variety of compds. with the  $\alpha$ -hydroxyketo linkage. G. M. Kosolapoff

STAKHEEVA-KAVERZNEVA, YE.D.

New data on the chemical composition of oxycellulose. 62  
E. D. Stakheeva-Kaverzneva. *Doklady Akad. Nauk S.S.S.R.* 78, 481-3(1951); cf. *C.A.* 44, 1257e.—New data are presented as proof of the presence of ketone groups in the oxycellulose as well as indications of the position of these groups in the units of the cellulose macromol. Evidence is presented for the presence of highly unstable lactone bonds within the mol. The presence of such bonds is used to explain certain discrepancies in the literature on oxycellulose.  
J. Rovtar Leach

STAKHEYEVA-KAVERZNEVA, Ye.D.; IVANOV, V.I.; SALOVA, A.S.

Chemical transformations of the macromolecule of cellulose under the action of oxidizing agents. VI. The presence in oxycelluloses of the groupings of carbonic esters and their effect on the determination of uronic carboxylic groups. Bull. Acad. Sci. U.S.S.R., Div. Chem. Sci. '52, 199-204 [Engl. translation].  
(CA 47 no.19:10220 '53)



STAKHEEVA-KAVERZNEVA, Ye.D.

Chemical Abst  
Vol. 48 No. 9  
May 10, 1954  
Cellulose and Paper

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③  
Chemical transformations of the cellulose macromolecule caused under action of oxidizing agents. VII. Chemical transformations of cellulose during its oxidation by sodium hypochlorite. E. D. Stakheeva-Kaverzneva, V. I. Ivanov, and A. S. Salova. *Bull. Acad. Sci. U.S.S.R., Div. Chem. Sci.* 1952, 681-90 (Engl. translation).—See C.I. 47, 859i.  
J.C.L. H.

AF  
9-17-54

STAKHEYEVA-KAVERZNEVA, Ye.D.

Chemical transformation of the cellulose macromolecule under the action of oxidizing agents. VIII. Chemical changes of cellulose on oxidation with hydrogen peroxide. V. I. Ivanov, E. D. Stakheeva-Kaverzneva, and Z. I. Kuznetsova. *Bull. Acad. Sci. U.S.S.R., Div. Chem. Sci.* 1953, 341-50 (Engl. translation).—See *C.A.* 47, 10220d.

H. L. H.

(2)

Specific method of determination of carbonyl groups in oxycellulose E. D. Stakheeva-Kaverzneva and A. S. Salova (Inst. Org. Chem., Acad. Sci. U.S.S.R., Moscow). *Zhur. Anal. Khim.* 8, 365-9(1953).—The interference of lactones (which react with  $\text{NH}_2\text{OH}\cdot\text{HCl}$ ) is prevented in this method by using an alc. soln. instead of an aq. one and by lowering the pH to 3-4. To prep. the reagent dissolve 5 g. of  $\text{NH}_2\text{OH}\cdot\text{HCl}$  in 240 ml. of 96% EtOH to which are added 5 ml.  $\text{H}_2\text{O}$  and 6 ml. of 0.2% bromophenol blue soln. (indicator). To this soln. add dropwise 0.2N NaOH until the color is green in transmitted and red in reflected light. The pH of this reagent is 3.2-3.4. To 2-3 g. of oxycellulose add 60 ml. of reagent and allow to stand for 2 hrs. Filter and titrate 50 ml. of filtrate with 0.1N NaOH to the original color or preferably to pH 3.2. Calc. the CO groups (x) present from  $x = 0.3360a/\text{wt. of sample}$ , where  $a$  is the no. of ml. of 0.1N NaOH used in titration. Use fresh reagent for each detn. M. Hosh...

STAKHEYEVA-KAVERZNEVA, Ye.D.; IVANOV, V.I.; SALOVA, A.S.

Chemical transformations in cellulose under the action of hypochlorite.  
Dumash. Prom. 28, No.7, 6-11 '53. (MLRA 6:7)  
(CA 47 no.22:12799 '53)

3  
Hydrogen peroxide bleaching of cellulose. B. D. Stak-  
heeva-Kaverzheva, V. I. Ivanov, and Z. I. Kuznetsova.  
Tshisu. Prom. 14, no. 3, 81-4 (1954). Oxidation processes  
occurring in cellulose during  $H_2O_2$  bleaching are discussed  
in detail. The decrease in viscosity ( $\eta$ ) of bleached cellu-  
lose (detd. in cuprammonium) is dependent on the pH of the  
bleaching bath since this det. the nature and the degree of  
oxidation of the macromol. Neutral medium yields cellu-  
loses contg. more aldehydic, ketonic, and dialdehydic groups  
than the alk. medium; being unstable in strong alkali these  
groups will considerably lower  $\eta$ . Optimum pH for  $H_2O_2$   
bleaching is 8-10; a higher pH favors the enolization of  
hydroxy ketones and hydroxy aldehydes, resulting in a low  $\eta$ .  
Elisabeth Barabash

114  
7-20-51

STAKHEYEVA-KAVERZNEVA, Ye. D.

*Matts* ✓ Chemical transformations of cellulose macromolecules under influence of oxidizing agents. XI. Causes of chemical instability of celluloses oxidized by nitrogen dioxide. E. D. Stakheeva-Kaverzneva and S. A. Kist. *Bull. Acad. Sci. U.S.S.R., Div. Chem. Sci.* 1956, 803-13 (English translation).--See *C.A.* 50, 17426e. B. M. R. 2

STAKHEYEVA-KAVERZNEVA, Ye. D.

15  
 ✓ Chemical transformations of cellulose in activated hypochlorite oxidation. Ye. D. Stakheeva-Kaverzneva, V. I. Ivanov, and G. A. Krylova (N. D. Zelinskii Inst. Org. Chem., Moscow). *Izv. Akad. Nauk S.S.S.R., Otdel. Khim. Nauk* 1957, 120-2; cf. *C.A.* 47, 850i. — Oxidation of cellulose with NaOCl (0.1N at pH 11 at 20° 1-24 hrs.) in the presence of urea as activator (0.025 moles/mole NaOCl) has the same characteristics as does the unactivated oxidation with NaOCl (kinetic curves shown), but the degree of polymerization (mol. wt.) does not change quite so drastically. Possibly the activator serves to oxidize the colored satellites of cellulose rather than cellulose itself.

G. M. Kosolapoff

Rm  
 AM

3 4E2C1

1. STAKHEYKO, F. G.
2. USSR (600)
4. Elm
7. Sowing elm seeds with alae removed. Les.khoz. 6 No. 3, 1953.
9. Monthly List of Russian Accessions, Library of Congress, April 1953, Uncl.



CATEGORY : Forestry. Dendrology.

SER. JOUR : Vestn. Zool.-Biologiya, No. 5, 1959, No.20132

AUTHOR : Stukheyko, F.G.

TITLE : Siberian Larch in the Southern Part of the  
Ukraine.

ORIG. PUB : Kolgospnik Ukraini, 1958, No.4, 39

ABSTRACT : No abstract

STAKHEYKO, Fedor Grigor'yevich, kand. sel'khoz. nauk; PIS'MENNYI, N.R.,  
~~red.~~; CHUGUNOVA, Z.S., red. izd-va; SHIBKOVA, R.Ye., tekhn.  
red.

[Cultivation of Siberian larch] Razvedenie listvennitsy sibir-  
skoi. Moskva, Goslesbumizdat, 1962. 71 p. (MIRA 15:10)  
(Larch)

STAKH, A.M.

Classes of motions with periods in partially ordered dynamic systems.  
Dif. urav. 1 no.5:619-624 My '65. (MIRA 18:7)

1. Institut matematiki s Vychislitel'nyy tsentrom AN Moldavskoy SSR.

STAKHIRA, I.M. [Stakhyra, I.M.]

Growing of  $\text{In}_2\text{Se}$  single crystals by the method of extraction  
from the melt. Ukr. fiz. zhur. 8 no.9:970-974 S '63.

(MIRA 17:8)

1. Chernovitskiy gosudarstvennyy universitet.

BORETS, A.N. [Borets', O.M.]; STAKHIRA, I.M. [Stakhyra, I.M.]

Self-absorption edge in  $\text{In}_2\text{Se}$  single crystals. Ukr. fiz. zhur.  
8 no.9:1026-1027 S '63. (MIRA 17:8)

1. Chernovitskiy gosudarstvennyy universitet.

BORETS, A.N. [Borets', O.M.]; STAKHIRA, I.M. [Stakhyra, I.M.]

Optical properties of  $\text{In}_2\text{Se}$ . Ukr. fiz. zhur. 9 no. 10:1074-  
1078 O '64 (MIRA 18:1)

1. Chernovitskiy gosudarstvennyy universitet.

L 14616-66 EWT(1)/ENT(m)/EWG(m)/T/ENF(t)/EWP(b) IJP(c) RIN/JD/CG  
ACC NR: AT6002263 (A) SOURCE CODE: UR/2564/65/006/000/0284/0287

AUTHOR: Stakhira, I. M.

ORG: none

TITLE: Growing of In<sub>2</sub>Se single crystals by the Czochralski method [Paper presented at the  
Third Conference on Crystal Growing held in Moscow from 18 to 25 November, 1963]

SOURCE: AN SSSR. Institut kristallografii. Rost kristallov, v. 6, 1965, 284-287

TOPIC TAGS: indium compound, selenide, single crystal growing

ABSTRACT: A typical apparatus for growing single crystals by the Czochralski method was employed, but some of its parts were modified to correspond to the particular characteristics of the crystallization of In<sub>2</sub>Se (control by means of temperature conditions). The alloys used for growing the crystals had to be homogeneous in order to produce satisfactory results. Since the most important factor affecting the single-phase character of the In<sub>2</sub>Se crystals is the temperature gradient in the region of crystallization, the Peltier effect was used to achieve the maximum temperature gradients. The conditions of growth were chosen

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ACC NR: AT6002263

on the basis of results of differential thermal analysis. X-ray analysis confirmed the rhombic system of  $\text{In}_2\text{Se}$ . Optical methods established that  $\text{In}_2\text{Se}$  is an anisotropic semiconductor with forbidden zone widths of 0.68 - 0.69 eV in the cleavage plane along the principal directions. The free carrier concentration was estimated to be  $10^{14} \text{ cm}^{-3}$ . Orig. art. has: 3 figures. 0

SUB CODE: 20 / SUBM DATE: none / ORIG REF: 001 / OTH REF: 002

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Card 2/2



L 57550-65 EWT(1)/EWT(m)/EWG(m)/EEC(t)/EWP(t)/EWP(b) Pz-6 IJP(c) RDW/JD/AT

ACCESSION NR: AP5014579

UR/0181/65/007/006/1777/1782

AUTHOR: Zhad'ko, I. P.; Rashba, E. I.; Romanov, V. A.; Stakhira, I. M.;  
Tovstyuk, K. D.

TITLE: Anistropy of electric and photoelectric properties of In<sub>2</sub>Se

SOURCE: Fizika tverdogo tela, v. 7, no. 6, 1965, 1777-1782

TOPIC TAGS: Dember effect, transverse Dember effect, electron mobility, hole mobility, anisotropic semiconductor

ABSTRACT: A theory of the transverse Dember effect was derived on the basis of the difference in the anisotropy of electron and hole mobility in macroscopic anisotropic semiconductors. The theory attributes the emergences of nonequilibrium carriers at the specimen's rear side to the influence of the field of the transverse photoelectromotive force. In<sub>2</sub>Se specimens consisting of single crystals 1 to 3 cm<sup>3</sup> in volume were used to detect experimentally the transverse Dember effect. This material was selected because of its structural anisotropy (the lattice of In<sub>2</sub>Se is diamond-shaped (D<sub>2h</sub>) with parameters  $a = 4.065 \text{ \AA}$ ,  $b = 12.24 \text{ \AA}$ ,  $c = 15.23 \text{ \AA}$ ) and its photosensitivity. The type of conductivity of the specimens was determined from the sign of the thermal emf and the Hall effect, and from the sign of photo

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ACCESSION NR: AP5014579

emf during illumination of a point tungsten probe. Investigations were made of electron-type specimens with  $\rho \sim 10^{-1}$  and  $\rho \sim 10^3$  ohm.cm and hole-type specimens with  $\rho \sim 10^3$  ohm.cm. The electron mobility in low ohmic electron-type  $\text{In}_2\text{Se}$  specimens at room temperature was of the order of  $500 \text{ cm}^2/\text{v} \cdot \text{sec}$ . The hole mobility at room temperature was smaller by at least one order. During illumination of a plate cut out at a  $45^\circ$  angle to the a-axis and containing a b-axis (light beam intensity  $3 \times 10^{15}$  quanta/sec) a photo emf of several millivolts was observed in the direction perpendicular to axis b. A photo emf of the same order was also observed on specimens cut out at a  $45^\circ$  angle to the a-axis but containing a c-axis. No photo emf was observed during the illumination of faces perpendicular to the b-axis. The measured characteristic times for photoconductivity and transverse photo emf coincided and for various n- and p-type specimens were  $\tau = 0.1-0.4 \text{ } \mu\text{sec}$ . Thus, at such small  $\tau$  the contribution to the transverse effect by anisotropic thermal emf caused by the heating of specimens is negligible. Orig. art. has: 13 formulas and 2 figures. [JA]

ASSOCIATION: Institut poluprovodnikov AN UkrSSR, Kiev (Institute of Semiconductors, AN UkrSSR)

SUBMITTED: 08Jan65

NO REF SOV: 004

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ENCL: 00

OTHER: 003

SUB CODE: SS,EM

ATD PRESS: 4037

L 16368-65 EWT(m)/EWP(t)/EWP(b) IJP(c)/ESD(t)/SSD/AFWL/ASD(a)-5/AS(mp)-2/  
RAEM(a) RDW/JD  
ACCESSION NR: AP4048864 S/0185/64/009/010/1074/1078

AUTHOR: Borets', O. M. (Borets, A. N.); ~~Stakhira, I. M.~~ (Stakhira, I. M.)

TITLE: Optical properties of  $\text{In}_2\text{Se}$  B

SOURCE: Ukrayinsk'ky fizychny zhurnal., v. 9, no. 10, 1964, 1074-1078

TOPIC TAGS:  $\text{In}_2\text{Se}$ , optical property, crystal orientation, refractive index, indium selenide

ABSTRACT: The anisotropy of the optical properties of  $\text{In}_2\text{Se}$  single crystals was investigated. The orientation of the crystals was determined by x-ray analysis and by infrared polarized radiation, based on the anisotropy on the absorption edge. The spallation surfaces and polished surfaces of the samples were examined. A distinct interference pattern with beats was observed in unpolarized and polarized radiation where the electric vector did not coincide with any of the basic crystallographic directions. This was explained by the optical biaxiality of the  $\text{In}_2\text{Se}$  crystal. The two principal refractive indices were determined. A depend-

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ACCESSION NR: AP4048864

4

ence between the absorption coefficients and the photon energy at room temperature for the three basic crystallographic directions was noted. A characteristic shape of the absorption edge with a break in the curve was obtained for the two basic crystallographic directions in the spallation plane. There was no such break when the electric vector was directed along the normal to the spallation plane. Transmission curves of 90 micron thick samples are presented for the three orientations of the crystal with respect to the electric vector. Temperature measurements indicated a complex mechanism of the basic transitions, but the complexity of the absorption edges made it impossible to interpret the nature of the basic transitions. "The authors thank K. D. Tovstyuk and P. A. Kotsyumas for advice and interest in the work and T. S. Gertovy\*ch for assisting in the investigation." Orig. art. has: 7 figures and 1 equation

ASSOCIATION: Chernivets'ky\*y derzhuniversy\*tet (Chernivets State University)

SUBMITTED: 01Nov63

ENCL: 00

SUB CODE: GP, IC

NO REF SOV: 003

OTHER: 001

Card 2/2

TAMMILA Y. S.

Machine for rolling disk saws. Der. from. LA no. 2428 F 165.

(MIRA 18:6)

STAKHUYEV, Yu.M.; VARLANOV, Yu.F.

Plane saw designed by A.P. Den'kach. Der. prom. 13 no.12:  
28-29 D '64 (MIPA 18:2)

STAKHIYEV, Yu.M.

Removing bottom rolls of saw frames. Der.prom. 8 no.1:26  
Ja '59. (MIRA 12:1)

1. Engel'skiy lesopil'nyy zavod.  
(Sawmills)

STAKHIYEV, Yu.M.

Forming framed-saw teeth on the PV-5 rolling mill. Der.prom. 8  
no.3:20 Mr '59. (MIRA 12:4)

1. Engel'skiy lesosavod.  
(Rolling (Metalwork)) (Saws)



STAKHIYEV, Yuriy Mikhaylovich; GOLUBEVA, T.M., red.; SHILLING, V.A., red.  
izd-va; BELOGUROVA, I.A., tekhn. red.

[Finishing sawing of lumber with circular saws without widening the  
toothed rim] Chistovaia raspilovka drevesiny kruglymi pilami bez  
ushireniia rezhushchego ventsa. Leningrad, 1961. 20 p. (Lening-  
gradskii Dom nauchno-tekhnicheskoi propagandy. Obmen peredovym opy-  
tom. Seriya: Derevoobrabatyvaiushchaia promyshlennost', no.3)

(MIRA 14:11)

(Sawing)

STAKHNEVICH, V.L., inzh.; ZADUNAYSKIY, Ya.N., inzh.

Work of the institute "Giprotorfrazvedka" in the prospecting of  
peat resources and designing of peat enterprises. Zbor. st.po  
izuch. torf.fonda no.2:5-14 '57. (MIRA 11:8)

1. Institut "Giprotorfrazvedka."  
(Peat)

Orig. Copy : Microbiology. Microbial Pathogenic for Man and  
Animals. Anaerobic Bacilli  
ABST. JOUR. : RZBiol., No. 3 1959, No. 10201  
AUTHOR : Segal', M. S., Stakhnovskaya, G. K.  
INST. : ---  
TITLE : Seeding of Soils with Spores and with the Tetanus  
Bacillus and the Tetanus Morbidity  
ORIG. PUB. : Vsb.: Anaerobnyye infektsii. Kiev, Gosmedizdat  
UkrSSR, 1957, 85-86  
ABSTRACT : 250 samples of earth were investigated by  
a biologic test on mice. When spores and  
bacilli of tetanus were present in them the  
infected mice developed muscular rigidity,  
contractures and convulsions on the 1st-3rd day,  
and the animals died. The percentage of positive  
findings ranged from 40 to 88, and in certain  
regions amounted to 100. A high degree of  
seeding of the soil was found in the region of  
irrigation fields. The specificity of the disease  
was checked by the injection of 10 antitoxic units  
CARD: 1/3

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SHEYINDAUM, N.M.; CHERNYSHEVA, P.I.; KOVTUNOVA, N.Ya.; YAKHINIS, Z.E.; STAKHO,  
A.S.; PONOMAREVA, T.D.

Duration of the usefulness of sterile solutions prepared in the  
pharmacy. Apt. delo 11 no.1:55-56 Ja-F '62. (MIRA 15:4)

1. Apteka Sochinskoy gorodskoy bol'nitsy No.2 i bakteriologicheskaya  
Laboratoriya Sochinskoy sanitarno-epidemiologicheskoy stantsii.  
(SOLUTIONS (PHARMACY))

STAKROBIDDINOV, S. S.

"Poisonous plants of the Fergan Valley"

Tashken. 1951. 32 pages with illustrations.

SO: Vet., May 1952, Unclassified.

Publication of the Academy of Sciences, Uzbek SSR, 1951. In Uzbek Language.  
A pamphlet to assist livestock breeders.

STAKHORESKAYA, L.K.; TOKAREV, B.I.

Effect of hydroxymethylfuraldehyde on yeast. Mikrobiologiya 33  
no.6:1056-1060 N-D '64. (MIRA 18:4)

1. Gosudarstvennyy nauchno-issledovatel'skiy institut gidroliznoy  
i sul'fitno-spirovoy promyshlennosti, Leningrad.

STAKHORSKAYA, L.K.

Improving the biological quality of the hydrolyzates of lignin  
cellulose. *Gidroliz. i lesokhim.prom.* 18 no.1:6-7 '65. (MIRA 18:3)

1. Gosudarstvennyy nauchno-issledovatel'skiy institut gidroliznyy  
i sul'fitno-spirovoy promyshlennosti.

ARISTOVSKAYA, T.V.; STAKHOESKAYA, L.K.

Nitrogen nutrition of bacteria of the dysenteric group; author's abstract.  
Zhur.mikrobiol.epid.i immn. no.11:43 N '53. (MLBA 7:1)

1. Iz Leningradskogo instituta vaktsin i syvorotok.  
(Dysentery) (Nitrogen)



NIKOGOSYAN, I.Kh.; STAKHORSKAYA, L.K.

Study of the alkaloids of the cow parsnip *Heracleum diasectum*. Trudy  
Len. khim.-farm. inst. no.17:198-201 '64. (MIRA 18:1)

1. Kafedra farmakognozii i botaniki Leningradskogo khimiko-farmatsevticheskogo instituta.

STAKHORSKAYA, L.K.

Investigating some cow parsnip species used in popular medicine.  
Trudy Len. khim.-farm. inst. 12:275-283 '61. (MIRA 15:3)

1. Kafedra farmakognozii i botaniki Leningradskogo khimiko-  
farmatsevticheskogo instituta.

(COW PARSNIP)  
(PHARMACOGNOSY)

STAKHORSKAYA, L.K.

Colorimetric method for determining coumarin derivatives, Apt. delo  
11 no.2:44-46 Mr-Ap '62. (MIRA 15:5)

1. Leningradskiy khimiko-farmatsevticheskiy institut.  
(COUMARIN) (COLORIMETRY)

SEMUSHINA, T.N.; STAKHORSKAYA, L.K.; MONAKHOVA, N.I.

Utilization of various sugars by fodder yeast cultures.  
Mikrobiologiya 32 no.5:863-868 S-0'63 (MIRA 17:2)

1. Gosudarstvennyy nauchno-issledovatel'skiy institut gidro-  
liznoy i sul'fitno-spirovoy promyshlennosti, Leningrad.

STAKHORSKAYA, L.K.; TOKAREV, B.I.

Effect of the resinous substances of hydrolyzates on yeast  
propagation. *Gidroliz. i lesokhim.prom.* 17 no.1:14-15 '64.  
(MIRA 17:4)

1. Gosudarstvennyy nauchno-issledovatel'skiy institut gidroliznoy  
i sul'fitno-spirovoy promyshlennosti.

STAKHORSKAYA, N. I.

USSR/General Biology - Ecology and Hydrobiology.

B-5

Abs Jour : Ref Zhur - Biologiya, No 1, 1957, 246.

Author : A.R. Prendes' and N.I. Stakhorskaya

Inst : Odessa Institute

Title : Hydrobiological and Fishery Economic Characteristics of Ponds in the Forestry Steppe and the Steppe Rayons of Odessa.

Orig Pub : Tr. Odessk. Un-ta, 1955, 145, 155-162.

Abst : Most of the ponds in the forestry-steppe rayons belong to the pond type, and the steppe rayons to the swamp type. Their hydrochemical characteristics, and information on the macrophytes, zooplankton, and "zoobentos" are given. The materials gathered in regard to the feeding of carp with data on a feed base lead to the conclusion of the necessity of taking measures to fertilize the ponds and provide food for the fish. An increase in fish production in the carp fisheries of Odessa Oblast may be achieved by the additional planting of tench and silver Prussian carp.

Card 1/1

PRENDEL', A.R. [Prendel', O.R.], prof.; STAKHORSKAYA, N.I. [Stakhors'ka, N.I.]

Hydrobiological and piscicultural evaluation of collective farm  
ponds in Odessa Province and outlook for increasing their fish  
yields. Pratsi Od. un. Ser.biol.nauk no.8(vol.147):115-121 '57.  
(MIRA 12:4)

(Odessa Province—Fish ponds)

PRENDEL', A.R. [Prendel', O.R.], prof.; KORENCHEVSKAYA, G.O. [Korenchevs'ka, H.O.]; STAKHORSKAYA, N.I. [Stakhors'ka, N.I.]

Materials on a study of the fauna, ecology and biology of leeches inhabiting bottom-land waters in the lower Dniester Valley. Pratsi Od. un. Ser.biol.nauk no.8(vol.147):123-125 '57. (MIRA 12:4)  
(Dniester Valley---Leeches)



L 21050-45 EWT(G)/EWT(1)/EED(A)-2/ERA(4)/EWG(+)/EEO(1)/EEO-4/ERA(H) Pz-6/Pz-4/  
Pr-5/Pas-2/Pg-1/Pet/Pt-10/Pl-4/Pl-4 JHR/GW/WE  
ACCESSION NR: AR5004877 S/0058/64/000/011/H062/H063

SOURCE: Ref. zh. Fizika, Abs. 11Zh390

AUTHORS: Bel'kovich, O. I.; Stakhov, A. A.

TITLE: Investigation of leading fronts of signals in inclined meteor propagation

CITED SOURCE: Sb. Meteorn. rasprostr. radiovoln. No. 1, Kazan', Kazansk. un-t, 1963, 70-79

TOPIC TAGS: meteoric radio scatter, meteor observation, pulse rise

TRANSLATION: The authors made a statistical analysis of the leading fronts of signals at two frequencies (in the 40 and 60 Mcs bands) reflected from meteor trails, for inclined meteor propagation. A connection is obtained between the average duration of the front and the registration level. It is shown that the average duration of

Card

1/2

L 31055-65

ACCESSION NR: AR5004877

the signals reflected from meteor trails is proportional to  $\lambda^2$ , whereas the average duration of the leading front is proportional to  $\lambda^{1/2}$  and consequently, with decreasing wavelength, the percentage of the time belonging to the leading front of the signals, which is the signal portion useful for information transmission, will increase. Thus, whereas in the 40 Mcs band the leading fronts account for 8--10% of the duty factor, at 100 Mcs the contribution of the leading fronts to the duty factor increases to 40--50%.

SUB CODE: AA, EC

ENCL: 00

Card

2/2

MINGULIN, R.I.; STAKHOV, A.A.

Automatic testing of high-stability oscillators by standard 100  
kHz frequency. Izv. tekhn. nauch. ts. 9:47-48 8 '64. (MIRA 18.2)

L 17962-63

EPF(c)/BDS/EWT(1) AFFTC/ASD Pr-4 CG

ACCESSION NR: AP3003703

S/0048/63/027/007/0947/0948

AUTHOR: Zhernovoy, A.I.; Rukhin, A.B.; Stakhov, O.V. 59

TITLE: Acceleration of polarization of nuclei in nuclear resonance proton probe

SOURCE: AN SSSR, Izvestiya, Ser.fiz., v.27, no.7, 1963, 947-948

TOPIC TAGS: nuclear magnetic resonance, NMR probe, polarizer

ABSTRACT: The main function of the polarizing system used in conjunction with a flow-through type nuclear magnetic resonance probe is to provide the maximum possible value of the magnetization vector of the nuclei of the liquid flowing into the coil. The conditions for insuring better than 90% polarization and less than 10% magnetization loss on the way from the polarizer to the probe coil are given in the form of equations in terms of the polarizer volume, the tubing volume and the relaxation time of the nuclei. The significance of these factors is discussed, and it is concluded that to satisfy the above requirements the ratio of the effective volume of the polarizer filled with sand and iron filings to the total volume of the polarizer should be 1 to 6 or better. That is, use of a "well filled" polarizer makes it feasible to reduce the volume of the gap of the polarizing magnet

Card 1/2

L 17962-63

ACCESSION NR: AP3003703

and increase the effective field strength. Orig.art.has: 5 formulas.

ASSOCIATION: none

SUBMITTED: 00

DATE ACQ: 02Aug63

ENCL: 00

SUB CODE: SD, PH

NO REF SOV: 0000

OTHER: 000

Card 2/2

L 32832-65 EWT(1) IJP(c)

ACCESSION NR: AP5004542

S/0048/65/029/001/0164/0165

AUTHOR: Zhernovoy, A.I.; Stakhov, O.V.

TITLE: Stabilization of a weak inhomogeneous magnetic field by means of nuclear magnetic resonance /Report, 14th Annual Conference on Nuclear Physics held in Tbilisi 14-22 Feb 1964.7

SOURCE: AN SSSR. Izvestiya. Seriya fizicheskaya, v.29, no.1, 1965, 164-165

TOPIC TAGS: inhomogeneous magnetic field, weak magnetic field, magnetic field measurement, nuclear paramagnetic resonance

ABSTRACT: Difficulties are encountered in stabilizing weak inhomogeneous magnetic fields by the nuclear magnetic resonance method because the inhomogeneity of the field calls for the use of a small transducer, while its weakness requires the use of a large one. These difficulties can in part be obviated by the use of the nuclear magnetic resonance head with a flowing resonating liquid described elsewhere by A.I.Zhernovoy and others (Priboiy i tekhnika eksperimenta No.2,115;No.5,73,1958; Inzh.-fiz.zh.1,95,1958; Izv.AN SSSR,Ser.fiz.22,993,1958), for the volume of liquid in the field to be stabilized can be kept small while the volume in the polarizing

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L 32832-65

ACCESSION NR: AP5004542

field, which determines the signal to noise ratio, is made large. A new difficulty arises, however, the system tends to oscillate because of the time lag between the presence of the working liquid in the stabilized field and its appearance in the polarizing field where the signal is delivered. It is proposed to avoid such oscillation by employing fast negative feedback circuits. A system for stabilizing the magnetic field of a  $\pi/2$  spectrometer is described briefly. The proposed feedback circuits are not described. Smooth control of the stabilized field is to be achieved by employing the beat frequency between a variable audio oscillator and a high-frequency crystal-controlled oscillator as the reference frequency for the nuclear magnetic resonance head.

ASSOCIATION: none

SUBMITTED: 00/--Jan65

ENCL: 00

SUB CODE: NP,EM

NR REF SOV: 004

OTHER: 000

Card 2/2





L 44788-65  
ACCESSION NR: AP5011063

tector 5. The amplitude of the NMR signal provides information regarding the polarization of the liquid in the sensor. When depolarized liquid is present in the sensor, there is practically no signal; when polarized liquid begins to flow into the analyzer, an NMR signal is developed which is coupled to a shaper circuit where a positive voltage gradient is then produced. This gradient is in turn passed to an electronic key (gate) and connects the generator to the marker, at which moment depolarized liquid begins to flow from the coil. If  $\lambda_0$  is the distance between the marker and the analyzer, and  $w$  is the rate of flow of the liquid, then within a time  $\Delta t = \frac{\lambda_0}{w}$  this depolarized liquid reaches the analyzer and the

nuclear magnetic resonance signal disappears. Meanwhile, in the shaping circuit, a negative voltage gradient develops which, by means of circuit 7, disconnects the generator from the marker, causing the depolarization of the liquid to cease. Within a time period  $\Delta t$  after this, an NMR signal appears and the process is repeated. The repetition frequency  $\Omega = \frac{1}{2\Delta t} = \frac{w}{2\lambda_0}$  is proportional to the flow

velocity of the liquid. This frequency is transformed, by means of circuit 9, into a DC voltage which is coupled to an arrow-type indicator and recorded. In

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L 44783-65

ACCESSION NR: AP5011063

practice, the device consists of two units: a remote sensor unit which is set up directly at the point at which the flow rate is to be measured, and an electronic unit which is connected to the remote unit by means of a 10-meter cable. Further details, technical specifications of components and dimensions are discussed in the article. The device is designed to measure the flow rate of water, acetone, and certain other liquids. The mean-square error, computed on the basis of point deviations from a calibration curve, does not exceed 1%. The test limits of the device run from 0.08 to  $1.4 \cdot 10^{-5} \text{ m}^3/\text{sec}$ . This equipment can be used to measure the flow-rates of liquids which provide a nuclear resonance signal and which have a nuclear relaxation time of not less than 0.1 sec. Orig. art. has: 2 figures.

ASSOCIATION: None

SUBMITTED: 00

ENCL: 01

SUB CODE: EC, MP

NO REF SOV: 002

OTHER: 000

Card 3/4

L 44788-65

ACCESSION NR: AP5011063

ENCLOSURE: 01

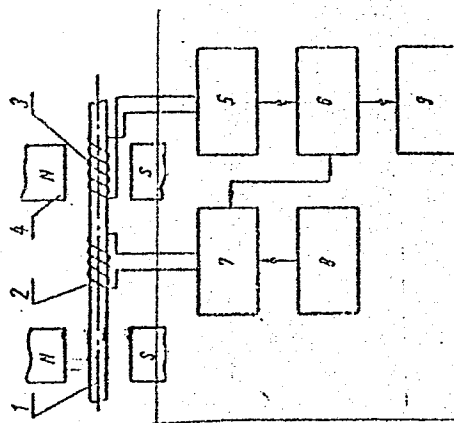


Figure 1. Block diagram of a pulse-frequency NMR flow-meter: 1 - strong permanent magnet, 2 - RF coil, 3 - circuit coil, 4 - magnet, 5 - detector, 6 - shaping circuit, 7 - electronic key (gate), 8 - generator, 9 - circuit for conversion of repetition frequency into DC voltage.

Card 4/4

YEKATERININ, V.V. ; ZHERNOVOY, A.I.; STAKHOV, O.V.

The IAMR pulse-frequency flowmeter. Izv. tekhn. no.3:54-56 Mr '65.  
(MIRA 18:5)